#### UG/CBCS/B.Sc./Hons./6th Sem./Chemistry/CHEMDSE3/2021



### UNIVERSITY OF NORTH BENGAL

B.Sc. Honours 6th Semester Examination, 2021

## **DSE3-CHEMISTRY**

# POLYMER CHEMISTRY

Full Marks: 40

#### ASSIGNMENT

The figures in the margin indicate full marks. All symbols are of usual significance.

		Answer any <i>four</i> questions from the following	10×4=40
1.	(a)	What are polymers? How are they classified?	1+4
	(b)	What do you mean by thermosetting and thermoplastic polymers? Give examples.	3
	(c)	Write down the structure of the monomers of PAN and PTFE.	2
2.	(a)	Define polymerization. Differentiate between addition and condensation polymerization.	2+3
	(b)	Explain the mechanism of cationic polymerization taking isobutylene as an example.	3
	(c)	Distinguish between inhibitors and retarders in radical polymerization with examples.	2
3.	(a)	Explain the kinetics of free radical chain growth polymerization and derive the equation for the rate of polymerization.	5
	(b)	Explain gelation and gel point in condensation polymerization.	3
	(c)	What is ring opening polymerization?	2
4.	(a)	What are copolymers and how are they classified? Derive the copolymer equation.	2+3
	(b)	What are Ziegler Natta catalysts? Discuss the mechanism of coordination polymerization of alkenes by Ziegler Natta catalysts.	5
5.	(a)	Explain number-average, weight-average and viscosity-average molecular weights of polymers with proper equations.	2+2+2
	(b)	Describe briefly how viscosity-average molecular weight of a polymer is determined by viscometry.	4
6.	(a)	Give the structure of PMMA. Discuss its properties and applications.	1+2+2
	(b)	Distinguish between the two types of phenol-formaldehyde resins.	3
	(c)	What are the major applications of poly (vinyl acetate)?	2
7.	(a)	How silicone is prepared? Discuss its properties and applications.	2+2+2
	(b)	What are polyurethanes?	2
	(c)	How Nylon 6, 10 is prepared?	2
8.	(a)	Discuss the preparation, properties and applications of Buna-S rubber.	2+2+2
	(b)	What is meant by vulcanization of rubber? Give examples of two vulcanizing agents.	2+2

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